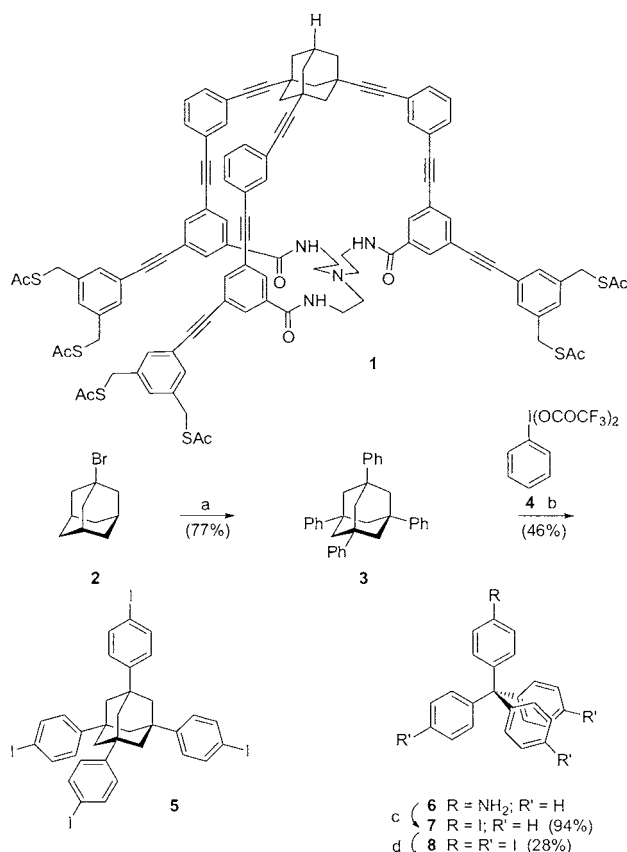


Scheme 1^a

^a Reaction conditions: (a) *t*-BuBr, AlCl₃, PhH, Δ 2 h; (b) **4**, I₂, (c) NaNO₂, HCl, then KI; (d) **4**, I₂.

tower-shaped molecule **1** (Scheme 1) as an inroad to a series of single molecule tips for AFM applications.^{10,11}

Herein, we describe the synthesis of several tetrahedrally shaped nanoscale molecules built around a 1,3,5,7-tetra-substituted adamantane unit or else a para-substituted tetraarylmethane core.¹² Given the high affinity that thiols,¹³ disulfides,¹³ and thiolacetates¹⁴ display toward gold, we included S atoms at the termini as a means to eventually anchor the molecular tip to a thin film of gold placed on a conventional AFM tip.¹⁵ We also demonstrate that **19** is

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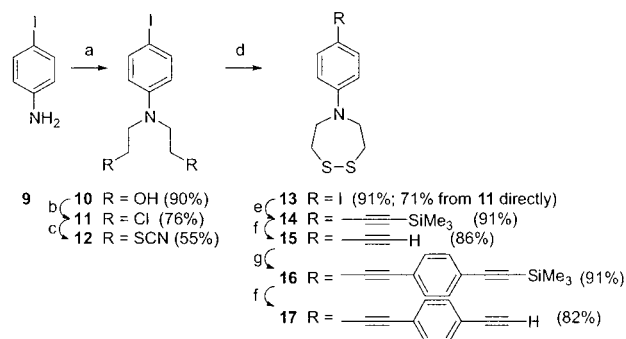
(13) The question of binding of disulfides and thiols to the gold surface is complicated by the thermodynamic stability of disulfide ring and monolayer formation kinetics. See the discussion in: Schlenoff, J. B.; Li, Ming; Ly, Hiep. *J. Am. Chem. Soc.* **1995**, 117, 12528.

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sufficiently rigid and robust to be visualized using a conventional AFM tip while laying on an atomically flat mica surface.

Tetrahedrally shaped precursors **5**¹⁶ and **8**¹⁷ were synthesized as follows. Tetraphenyladamantane **3** was prepared by arylation of 1-bromoadamantane in a mixture of hot benzene, *tert*-butyl bromide, and AlCl₃. The low solubility of **3** allows facile purification by washing with CHCl₃ in a Soxhlet extractor. Treatment of adamantane **3** with [bis(trifluoroacetoxy)iodo]benzene (**4**) and iodine gave tetraiodide **5**. Tetraiodide **8** was synthesized from *p*-tritylaniline (**6**), which was diazotized and reacted with KI to give iodide **7**. Further iodination of **7** with **4** in the presence of iodine gave tetraiodide **8**.

We chose a substituted [1,2,5]-dithiazepane unit for eventual binding of the tips to a gold surface. 4-Iodoaniline (**9**) was treated with ethylene oxide to give diol **10**, which was converted into dichloride **11** (Scheme 2).

Scheme 2^a

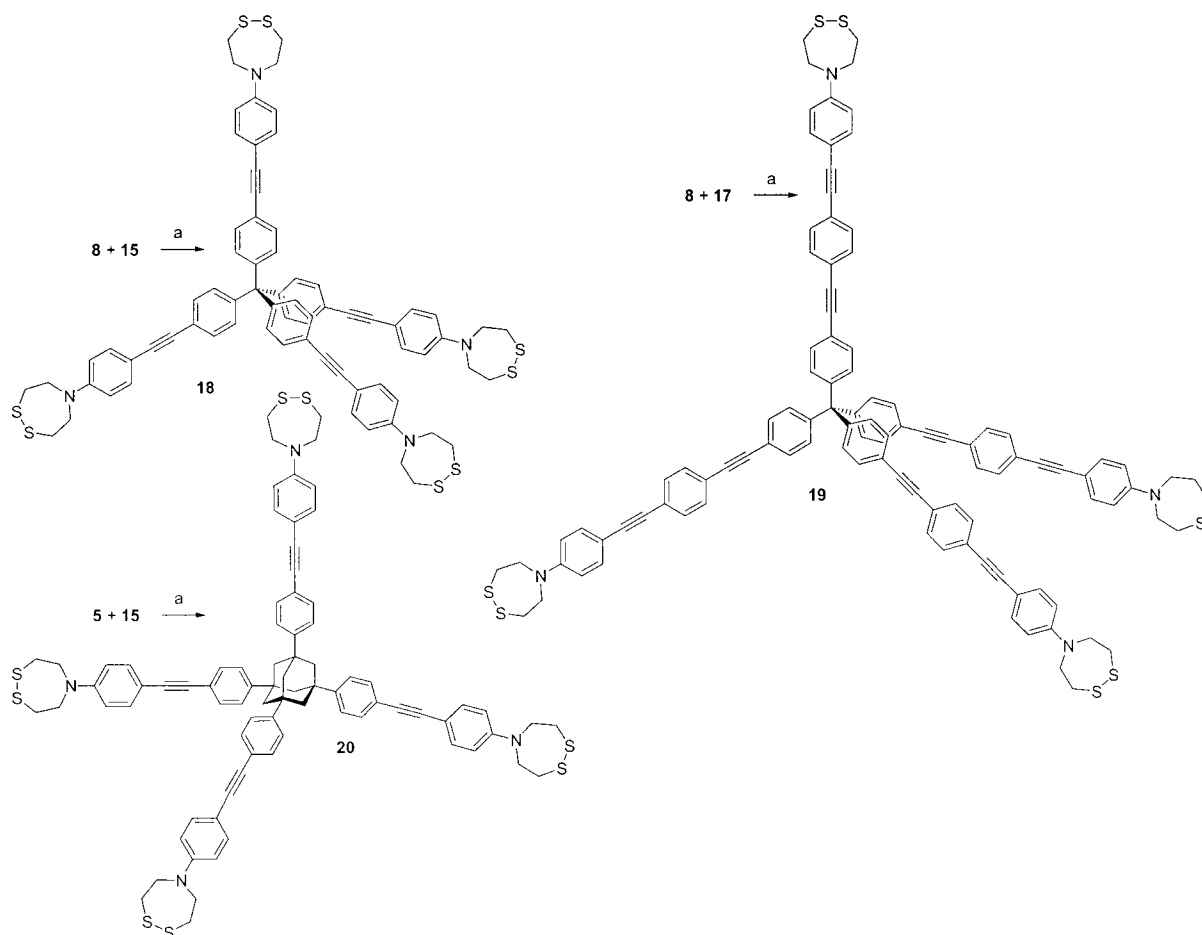
^a Reaction conditions: (a) Ethylene oxide, MeOH; (b) POCl₃, pyridine; (c) KSCN; (d) KOH; (e) TMSA, Pd(PPh₃)₂, CuI, THF/Et₃N; (f) K₂CO₃, MeOH; (g) (4-iodophenylethynyl)trimethylsilane, Pd(PPh₃)₂, CuI, THF/Et₃N.

Reaction of **11** with KSCN gave dithiocyanate **12**, which was converted into [1,2,5]-dithiazepane **13** upon treatment with hot KOH. Alternatively, reaction of **11** with excess KSCN afforded **13** directly. Reaction of **13** with trimethylsilylacetylene (TMSA) gave **14**. Cleavage of **14** gave alkyne **15**, which in turn was treated with (4-iodo-phenylethynyl)-trimethylsilane under Sonogashira coupling conditions to give TMS derivative **16**. Cleavage of **16** gave terminal alkyne **17**. Alkynes **15** and **17** were separately coupled to tetraiodide **8** under Sonogashira conditions to give tetraphenylmethane **18** and its larger analogue **19**, respectively (Scheme 3). Alternatively, tetraiodide **5** was coupled to alkyne **15** to give symmetrically tetrasubstituted adamantane **20**.

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Scheme 3^a

^a Reaction conditions: (a) Pd(PPh₃)₂, CuI, THF/Et₃N.

Visualization of Molecule 19 Using a Conventional AFM Tip. Freshly cleaved mica was separately spin-coated with 0.06, 0.25, and 1.00 μ M solutions of molecule **19** in CH₂Cl₂ and then examined by AFM in air and ambient humidity using a conventional tip with tapping mode (Figure 1B–D). The light spots are images of individual molecules¹⁸ laying on the mica surface. The observed heights¹⁹ of the molecules in Figures 1B–D are 0.8 ± 0.1 nm, 1.0 ± 0.2 nm, and 0.5 ± 0.1 nm.²⁰ The variation in height likely stems from day to day differences in humidity²¹ and set up parameters (drive amplitude A_0 and set point amplitude A_{sp} ^{22,23}). The height determined by AFM is lower than the calculated height based on molecular modeling. This is

consistent with similar differences observed for DNA²⁴ and C₆₀¹⁸ and may be due to molecular deformations caused by the AFM tip. As a control, freshly cleaved mica was separately spin-coated with the CH₂Cl₂ used as the solvent (Figure 1A) and a 1 μ M solution of segment **17** (Figure 1E) in CH₂Cl₂ and then examined by AFM. Segment **17** is much smaller than **19** and can lay flat on the mica surface. As expected, only the mica surface is seen in both control images, absent features analogous to those observed with **19**.

Figure 1F shows images of **19** in the presence of DNA fragments, which were deposited onto mica from aqueous buffer and washed extensively prior to spin-coating a 0.25 μ M solution of **19** in CH₂Cl₂. The observed height (0.6 ± 0.1 nm) of **19** is similar to that of the DNA fragments (0.7 ± 0.1 nm), as expected. Images B–D demonstrate that the number of individual molecules of **19** appearing in the images is concentration dependent.

(18) Individual fullerene molecules (C₆₀) on a Si(111) surface have been imaged by AFM in a noncontact mode in an ultrahigh vacuum. See: Kobayashi, K.; Yamada, H.; Horiuchi, T.; Matsushige, K. *Appl. Surf. Sci.* **1999**, *140*, 281.

(19) For a detailed discussion of resolution in AFM see: Bustamante, C.; Keller, D. *Physics Today* **1995**, *48*, 32.

(20) Confidence limits in all cases were calculated at a 75–80% probability level.

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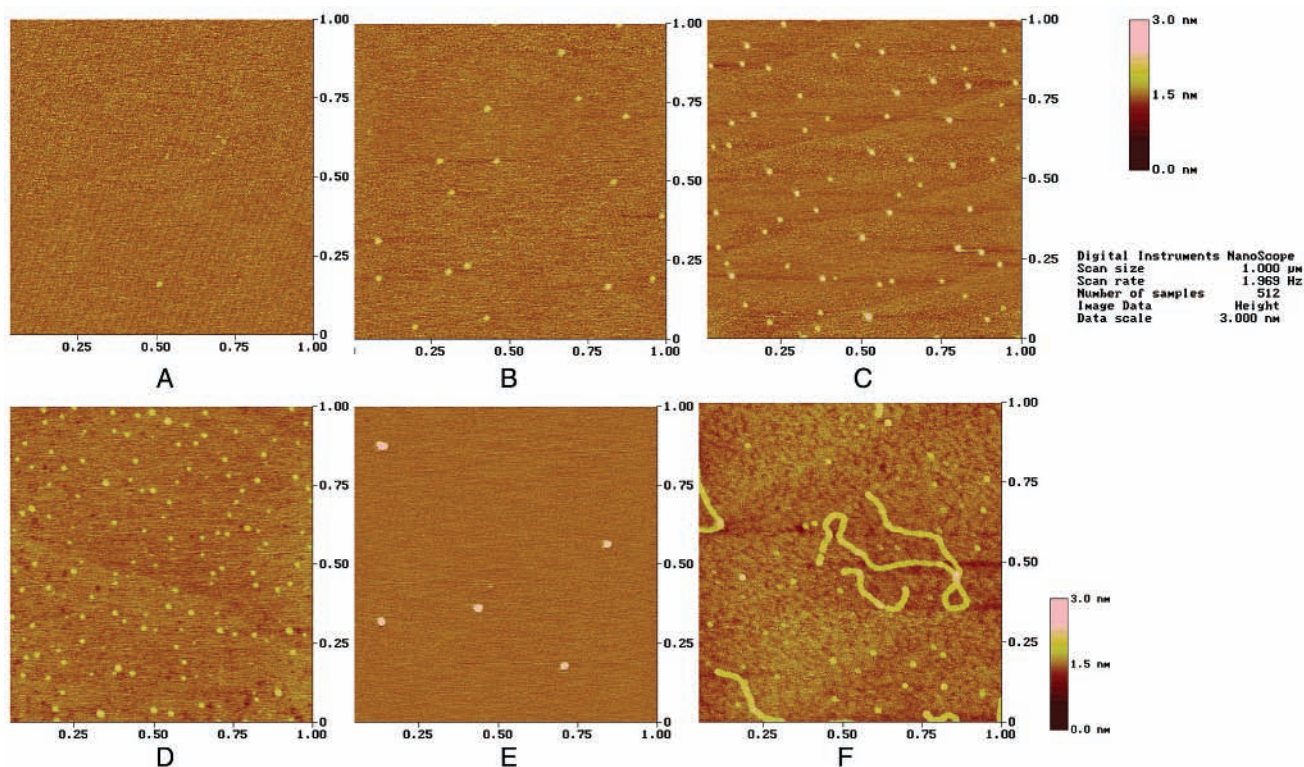


Figure 1. AFM images ($1.00 \times 1.00 \mu\text{m}$) of freshly cleaved mica after spin-coating with the following: A, CH_2Cl_2 used as the solvent (control image); B, $0.06 \mu\text{M}$ solution of **19** in CH_2Cl_2 ; C, $0.25 \mu\text{M}$ solution of **19** in CH_2Cl_2 ; D, $1 \mu\text{M}$ solution of **19** in CH_2Cl_2 ; E, $1 \mu\text{M}$ solution of segment **17** in CH_2Cl_2 (control image). For image F, DNA was applied to the mica prior to spin-coating a $0.25 \mu\text{M}$ solution of **19** in CH_2Cl_2 . Note the dependence of molecule **19** image density on concentration (B–D). Nanofeatures such as those in control images A (observed height, 0.6 nm) and E (observed height, $2.1 \pm 0.1 \text{ nm}$) are occasionally seen on certain regions of the surface when imaging freshly cleaved mica and may represent mica particulates produced as a result of the cleavage.²³

In summary, we have synthesized three tetrahedrally shaped nanoscale molecules **18–20**. These molecules are designed to serve as molecular tips for AFM and may also find application as chemically well-defined objects for calibration of commercial AFM tips. AFM images of **19** demonstrate that the molecule is sufficiently large and rigid to be imaged by a conventional AFM tip. Efforts to attach the molecules to commercial AFM tips are underway.

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Supporting Information Available: Experimental procedures and AFM conditions, ^1H NMR spectra for **5**, **7**, **8**, and **10–20**, and ^{13}C NMR spectra for **5** and **10–20**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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